



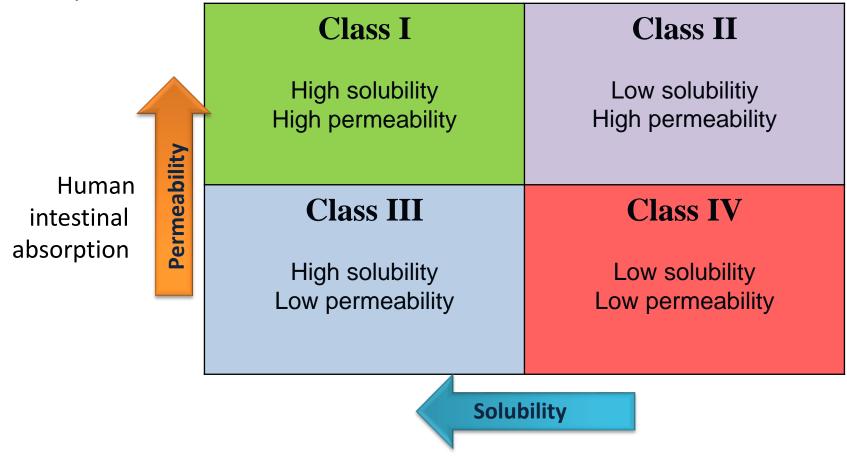


Gordon L. Amidon,

Amidon et al. published the importance of Biopharmaceutical Clasification System (BCS) in 1995, which is an important characteristic of the preparation's bioavailability.

According to their suggestion APIs can be categorized into four groups regarding the gastrointestinal absorption by their solubility and permeability characters.

APIs can be categorized into four groups according to their solubility and permeability:



Volume of water required to dissolve the highest dose streight across the physilogical pH range (1.2; 4.5; 6.8)

APIs with high solubility and high permeability

API absorbs well (though the metabolism can decrease the bioavailability), the rate controlling step will be the dissolution.

$$k_{diss} << k_{abs}$$

APIs with high solubility and high permeability

$$CH_3$$
— C — NH — OH

acetaminophen, paracetamol

propranolol

verapamil

imipramine

APIs with high solubility and high permeability

metronidazole

pindolol

H_3C N N CH_3

theophylline

salicylic acid

ephedrine

APIs with low solubility and high permeability

API's absorption is restricted by the dissolution's speed and amount.

Because of the low solubility the absorption occurs at the GI tract's longer regions, the dissolution lasts for a longer period of time and it influences the concentration at the place of absorption.

$$k_{diss} \le k_{abs}$$

APIs with low solubility and high permeability

diclofenac

$$CH_3$$
 CH_3
 CH_3
 CH_3

indometacin

ibuprofen

griseofulvine

APIs with low solubility and high permeability

APIs with high solubility and low permeability

The transfer speed through the membranes is very important in this case.

Long contact time with the mucus is required to increase the absorption (mucoadhesive systems).

$$k_{diss} >> k_{abs}$$

APIs with high solubility and low permeability

APIs with high solubility and low permeability

$$\begin{array}{c|c}
O & O \\
H_2N - S & NH \\
O & N \\
O & H
\end{array}$$

hydrochlorothiazide

doxazosin

cimetidine

APIs with high solubility and low permeability

BCS Class IV

APIs with low solubility and low permeability

Absoprtion of APIs belongign into this group is low, thus peroral administration is doubtful.

Technologies increasing the solubility should be applied: molecular dispersion, solubilisation.

BCS Class IV

APIs with low solubility and low permeability

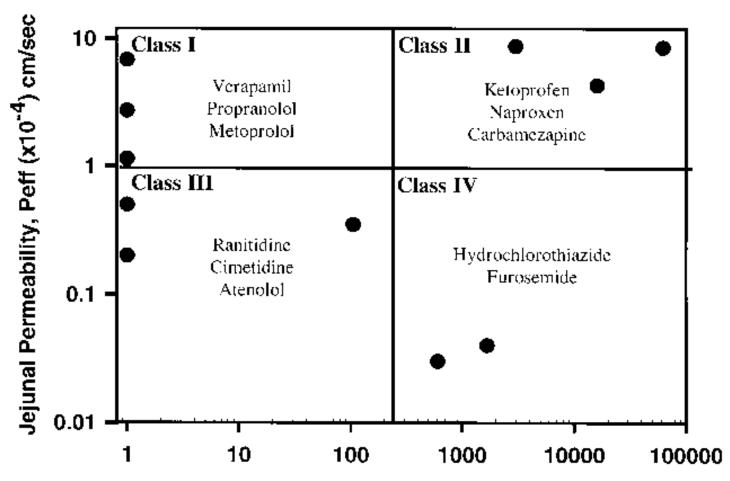
amphotericin B

BCS Class IV

APIs with low solubility and low permeability

hidrochlorotiazide

BCS Class Membership of Selected Model Drugs



Volume (ml) of required to dissolve the highest dose strength at the lowest solubility in pH 1-8 range (Note: 1 ml set as low value)

Three type of rates can be introduced according to the BCS that help researchers to develop preparations:

- absorption rate (A_n)
- dose rate (D_o)
- dissolution rate (D_n)

Absorption rate (A_n) – ration of residence time in the small intestine and the time needed for absorption:

$$A_n = \frac{t_{res}}{t_{abs}}$$

 t_{res} residence time t_{abs} time needed for absorption

<u>Dose ratio</u> (D_o) – can be calculated by the solubility, dose and tha amount (volume) of water used for the administration:

$$D_o = \frac{D}{V_o C_s}$$

D dose,

 V_o volume of water used at the administration, C_s solubility.

<u>Dissolution ratio</u> (D_n) – is the ratio of API's residence time in the GI tract and the time of dissolution:

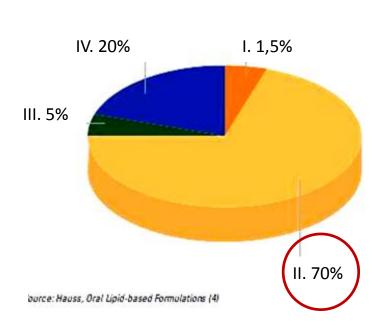
$$D_n = \frac{t_{res}}{t_{diss}}$$

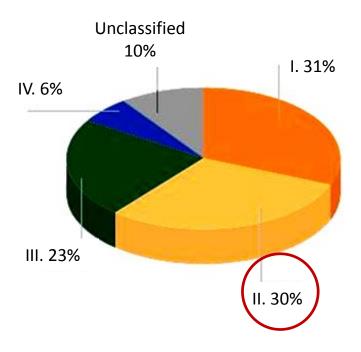
 t_{res} residence time, t_{diss} dissolution time,

Literature shows that there is an increasing number of poorly soluble compounds in the drug discovery pipeline – including 70 per cent BCS Class II compounds – although only about 30 per cent are BCS Class II in the existing top 200 market

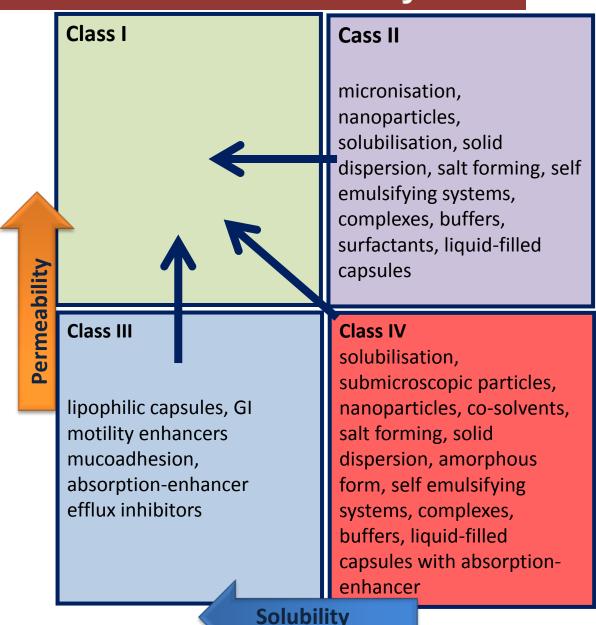
New chemical entities

Top 200 marketed drugs in USA





Sources of pharmaceutical technology to increase the bioavailability of drug delivery systems accourding to the BCS



Many THANKS for Your Attention

Dziękuję Ďakujem dhanya-waad Дякую bedankt ありがとう go raibh maith agat tesekkürle Спасибо Thank you Merci köszi tack så mycket Thank you faleminderit Shukriyâ Danke hvala díky kiitos takk Obrigada Mulţumesc nandri Ευχαριστώ Grazie anugurihiitosumi Muchas gracias köszönöm Terima Kasih 너를 감사하십시요 mange tak aitäh děkuji vam salamat