

ABSORPTION

University of Pécs

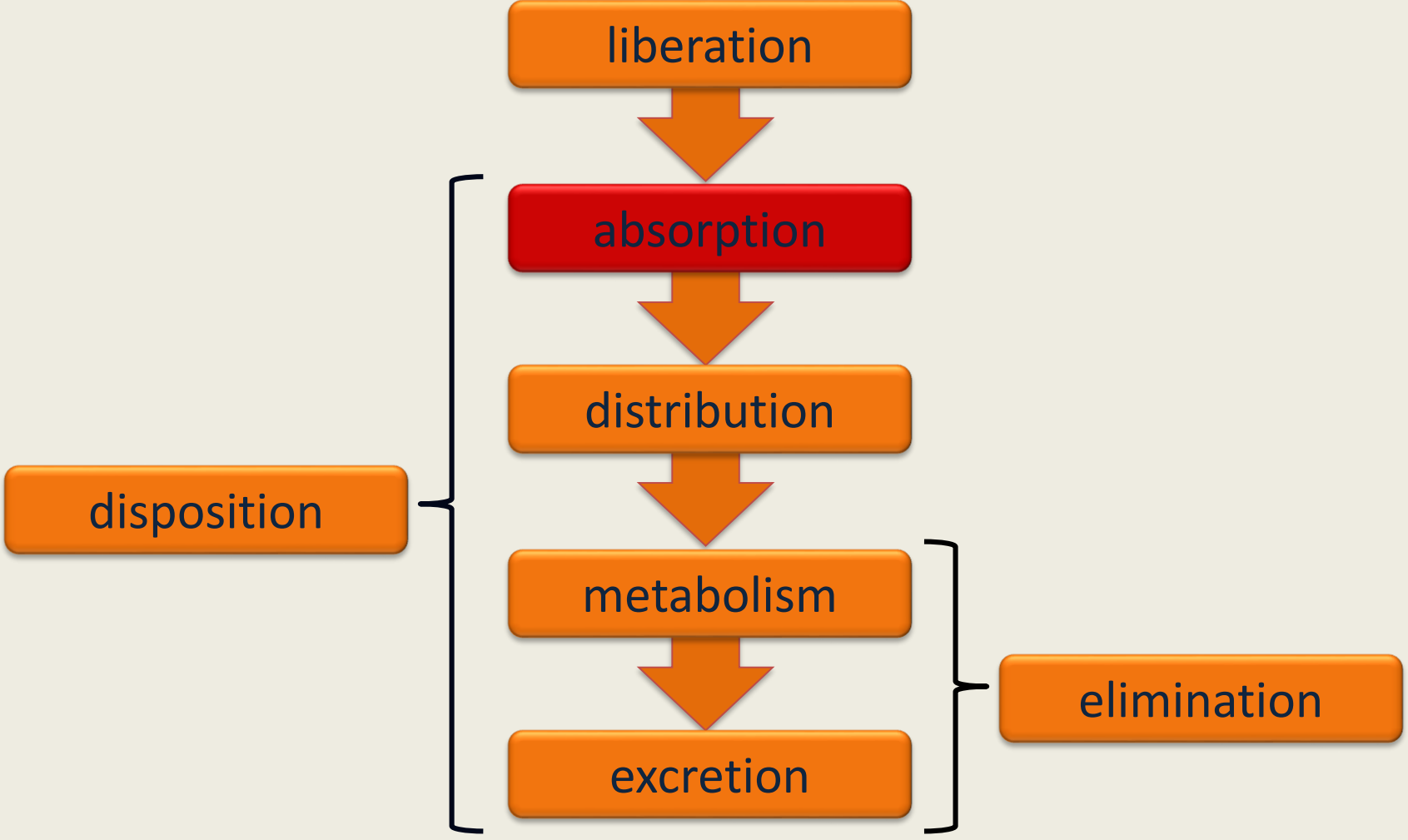
Institute of Pharmaceutical Technology and Biopharmacy

Absorption

Definition

Absorption is the movement of the API into the systemic circulation.

ADME → LADME



Transport processes

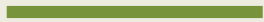
Intercellular and intracellular processes

API can be absorbed by

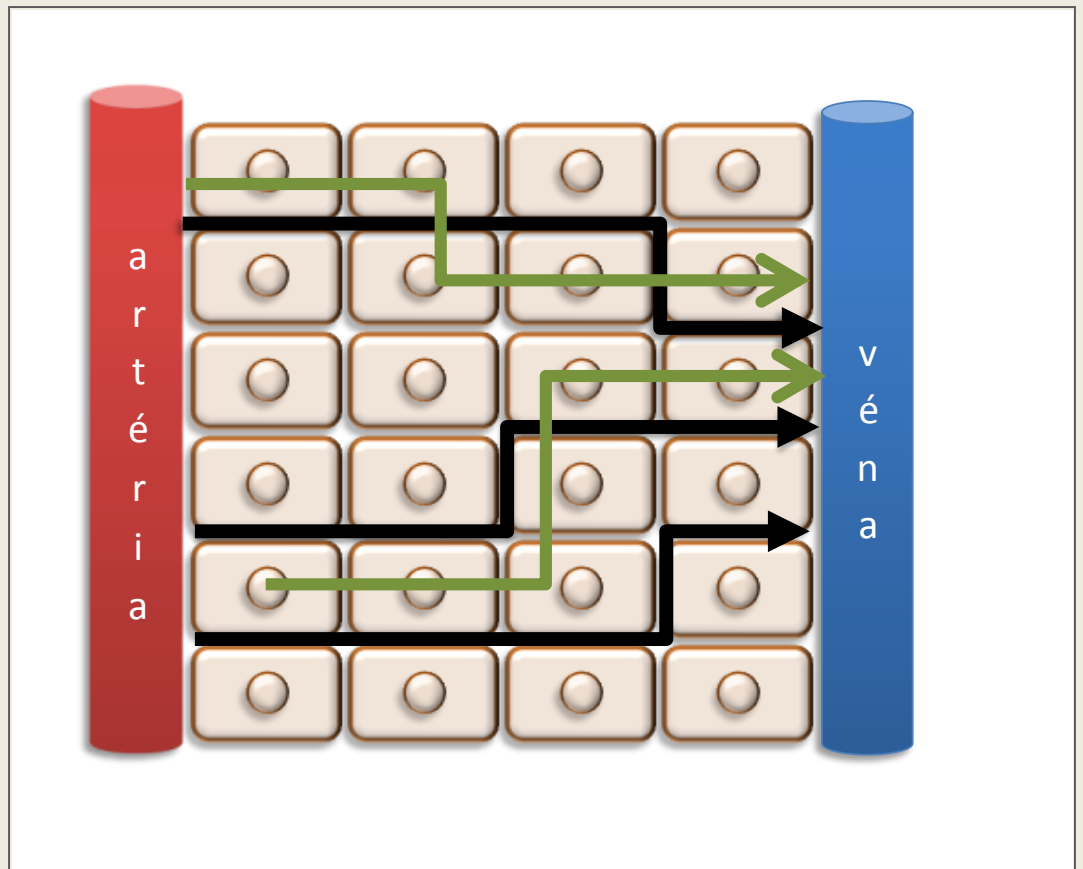
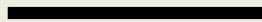
1. **transcellular** transport – across the cells
2. **paracellular** transport – through the gaps between the cells

Paracellular / transcellular transport

transcellular



paracellular



Transcellular/paracellular transport

Transcellular transport	<p>transport without carrier molecule</p> <ul style="list-style-type: none">- passive diffusion- ionpair transport <p>carrier-mediated transport</p> <ul style="list-style-type: none">- facilitated diffusion,- active transport <p>endocytosis (phagocytosis, pinocytosis)</p> <p>exocytosis</p>
Paracellular transport	filtration through pores

Paracellular transport

- Filtration through pores

It is determined by:

- Difference in pressure between the areas divided by the membrane
- Size of molecules
- Pore size

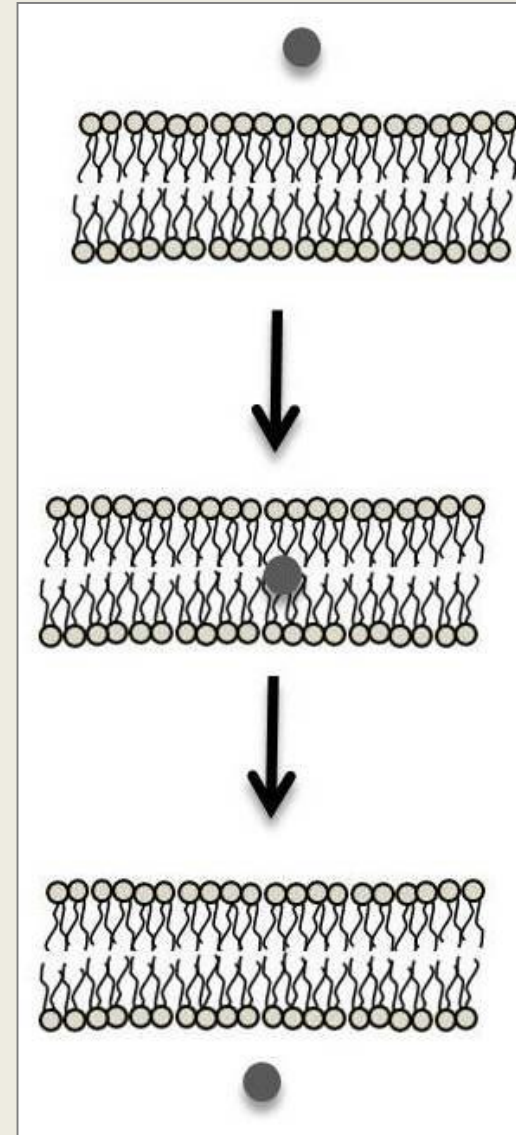
Substances: Ca ion, sugar, aminoacids, proteins

Passive diffusion

The most frequent type of transport of APIs.

Apolar, non-ionised APIs can cross the membrane according to the concentration gradient.

This type of transport lasts until the equilibrium in concentration, it does not need energy.



Ionisation

Degree of ionisation of molecules is calculated by the **Henderson-Hasselbalch** equation. Using this formula the ratio of ionised and non-ionised amount of molecules can be determined which highly depends on the pH of environment.

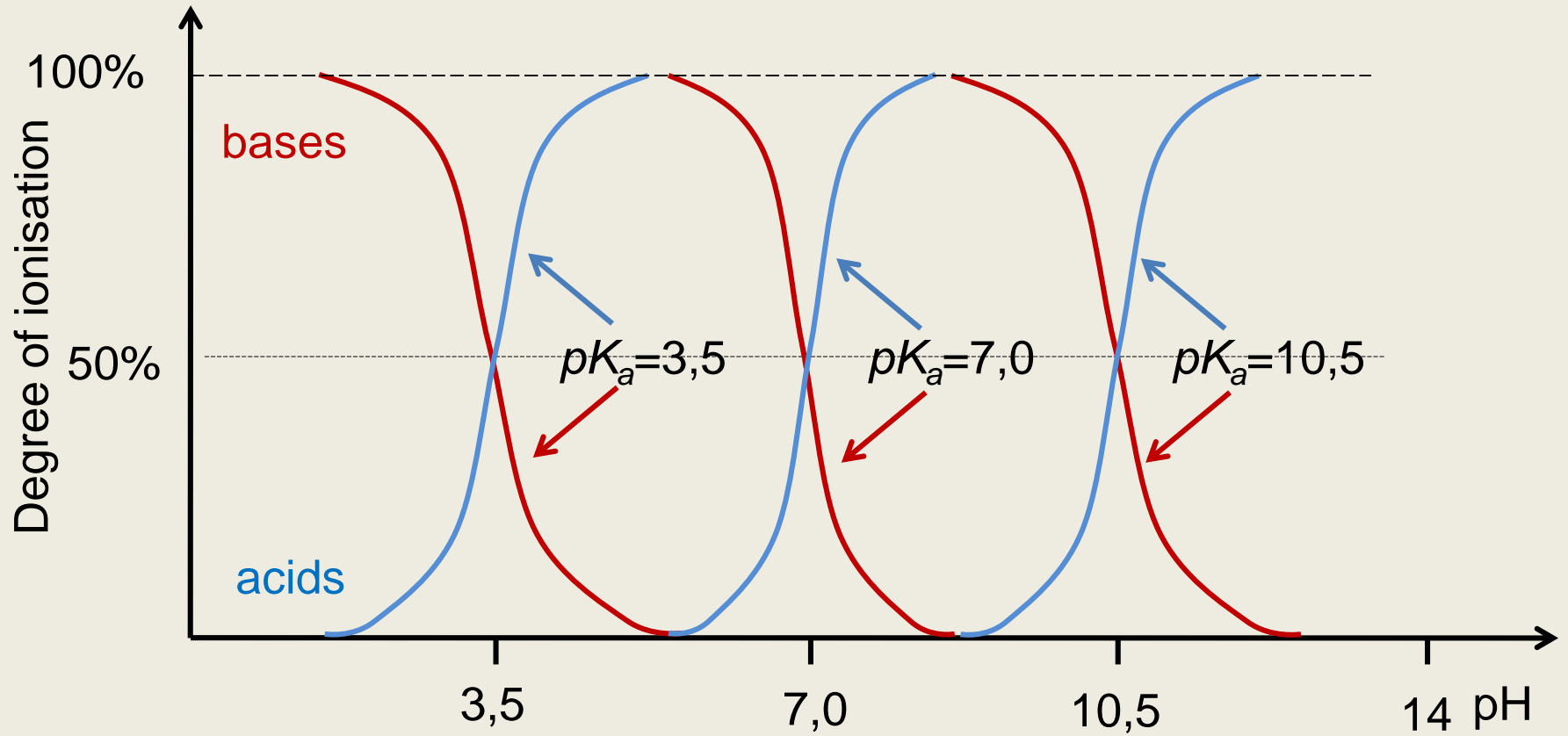
In case of weak bases:

$$pH - pK_a = \lg \frac{C_{non-ionised}}{C_{ionised}}$$

In case of weak acids:

$$pH - pK_a = \lg \frac{C_{ionised}}{C_{non-ionised}}$$

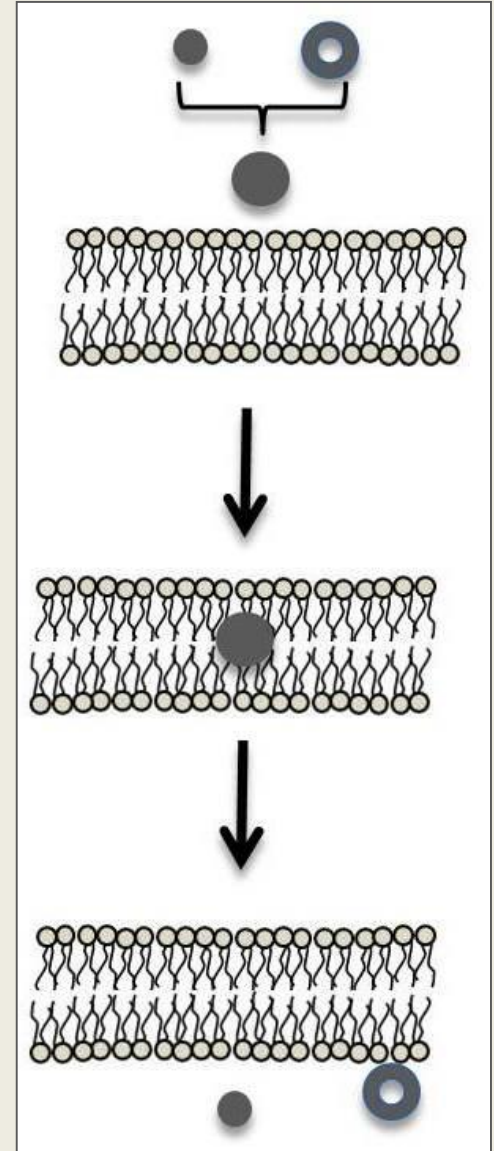
Ionisation



Ion-pair transport

Ion-pair transport facilitates the passive diffusion of ionised molecules with low lipid-water partition coefficient.

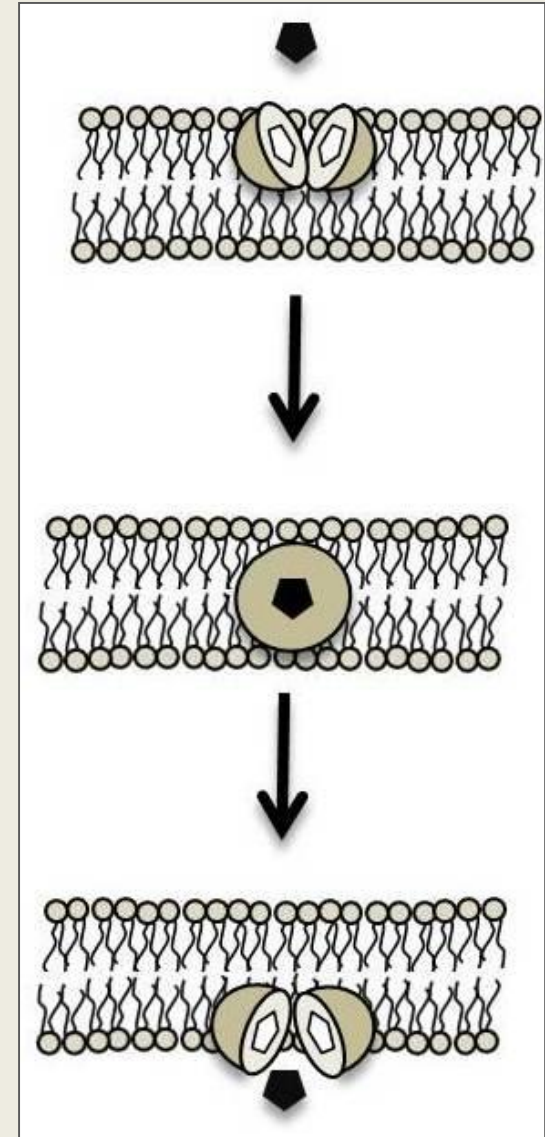
Lipophobic molecules make ion-pair complex with the GI tract's mucous or other molecules with opposite charge thus making an apolar, shielded complex which is able to cross the membrane by passive diffusion.



Facilitated transport

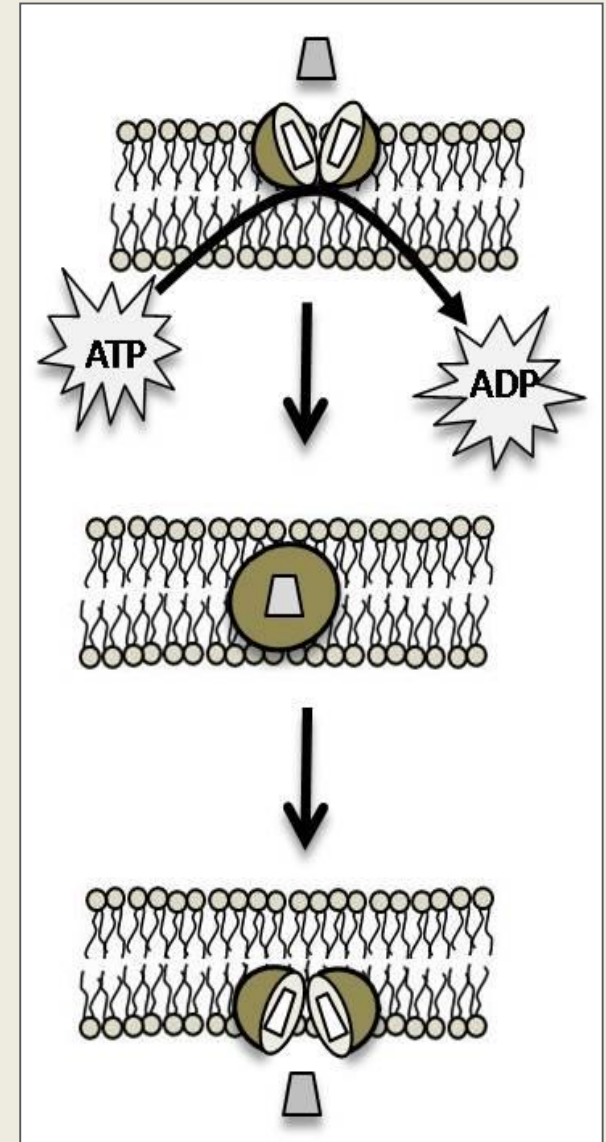
Mostly applicable at small, polar molecules.

Carrier molecule eases the crossing through the membrane by making a complex with the API. The transport mechanism is determined by the concentration gradient. It does not need energy, but it can be saturated.



Active transport

In this case the molecule can cross the membrane against the concentration gradient. This transport needs energy, which is provided by the ATP.



Endocytosis and exocytosis

Endocytosis – the cell absorbs molecules by engulfing them. Bigger particles (>500nm) are absorbed by **phagocytosis**, smaller particles are absorbed by **pinocytosis**.

Exocytosis is the opposite process to endocytosis.

DDS

A yellow rectangular button with rounded corners and a slight gradient, containing the text "DDS" in a bold, black, sans-serif font.

Absorption can be controlled by the liberation by making it the rate controlling step.

Speed of drug release

liberation

absorption

absorption

k

k_a

If $k > k_a$ then resultant rate = k_a

If $k = k_a$ then resultant rate = k_a

If $k < k_a$ then resultant rate = k

solution

solution

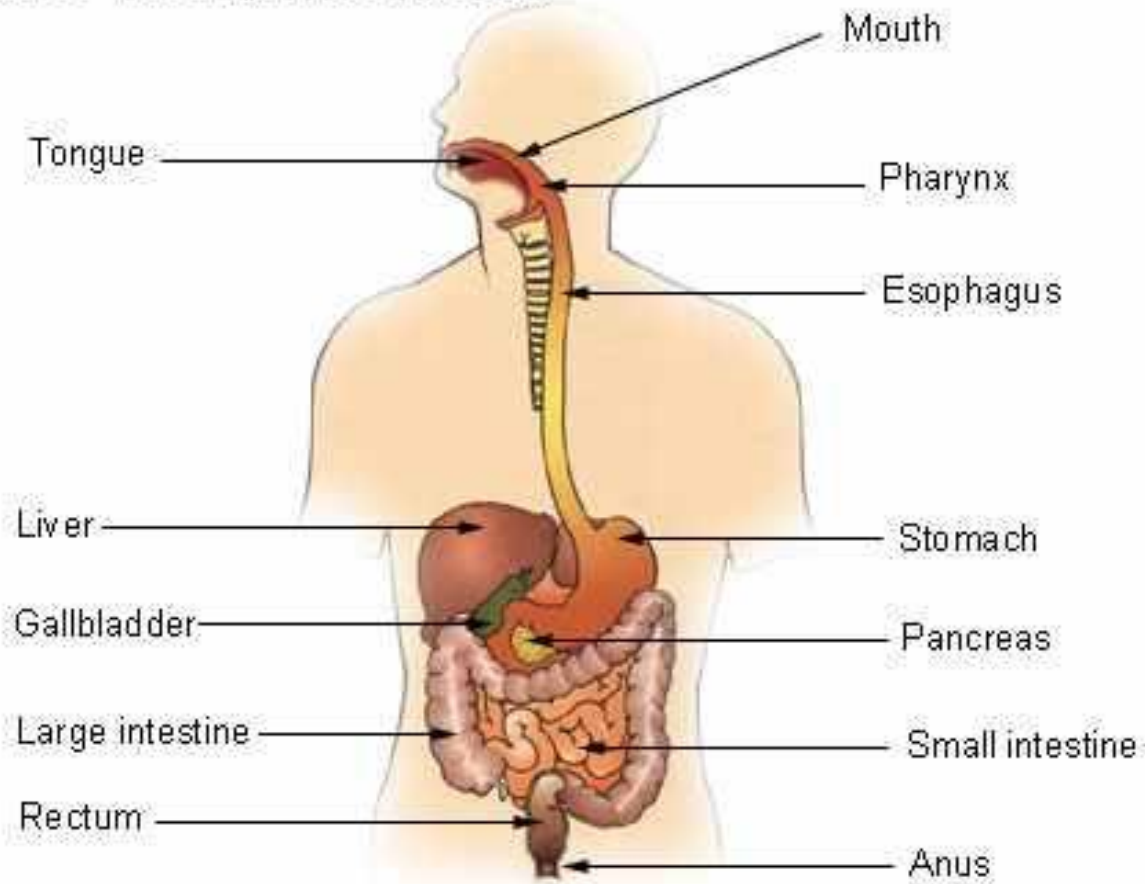
Routes of administration

Rate and speed of absorption is determined by the absorption surface (absorption window), which depends on the route of administration.

Absorption through the GI tract

GI tract

Organs of the Digestive System



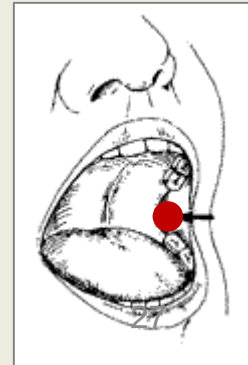
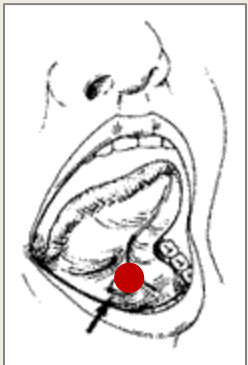
Characteristics of the GI tract

- huge absorption surface
- specific organs with specific tissue
- retention time
- specific pH environment
- specific enzymatic system

Mouth

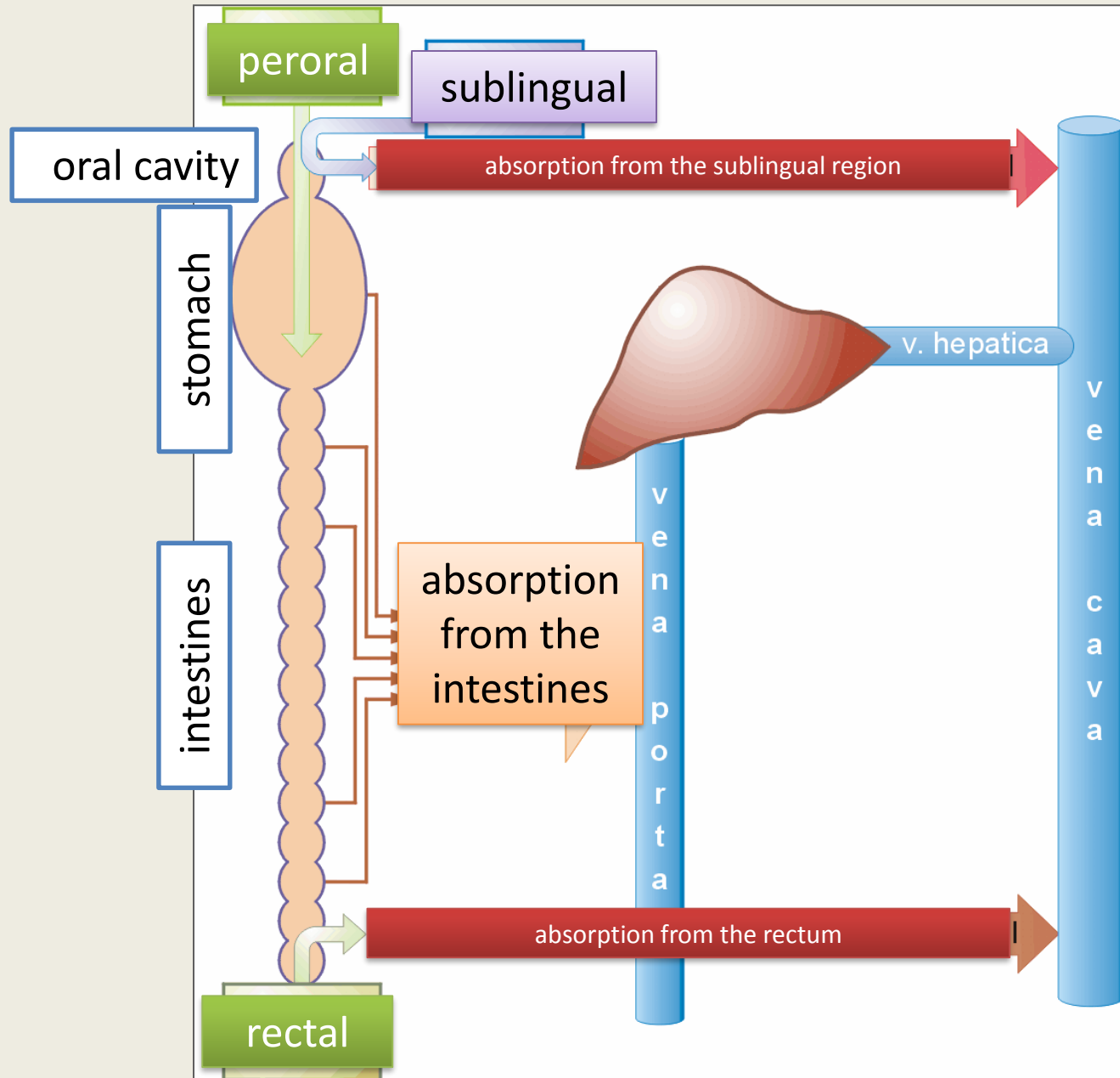


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Absorption

GI tract



nifedipin



nitroglycerol



captopril

Absorption from the sublingual region

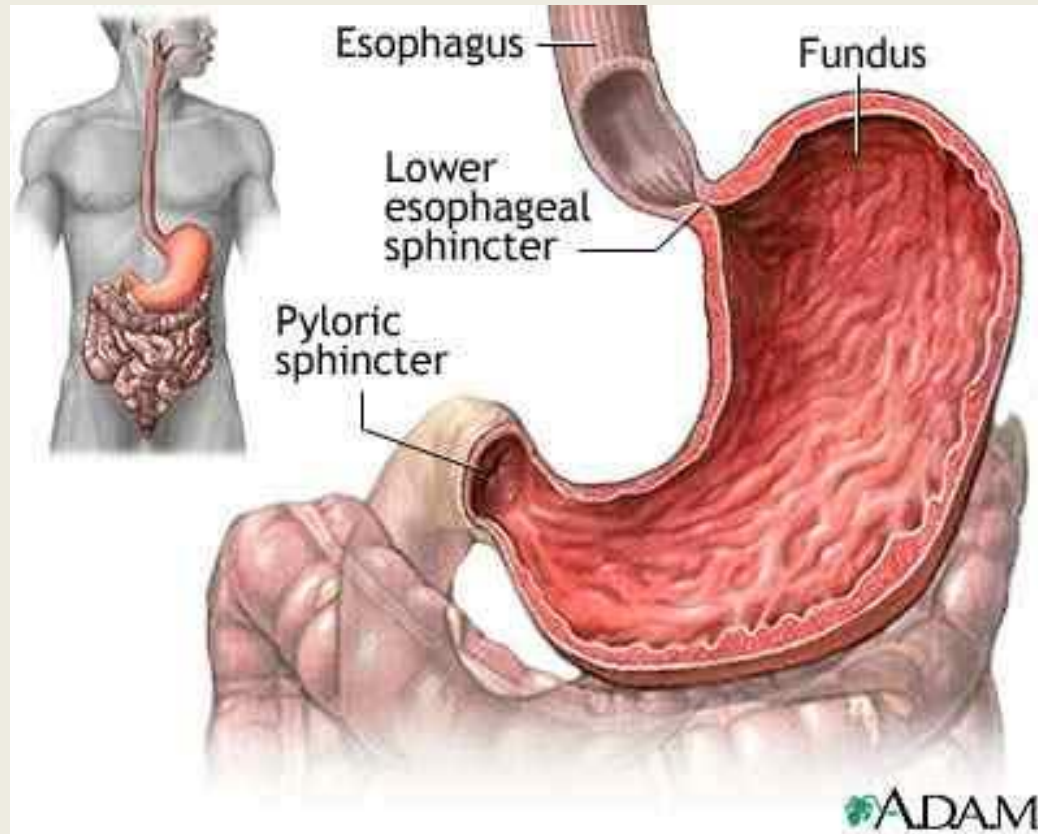
Advantage

- *Quick absorption*
- *No first-pass effect*
- *Increased patient compliance*

Disadvantage

- *May be irritative*
- *Unpleasant taste*
- *Cannot be administered to unconscious patients*

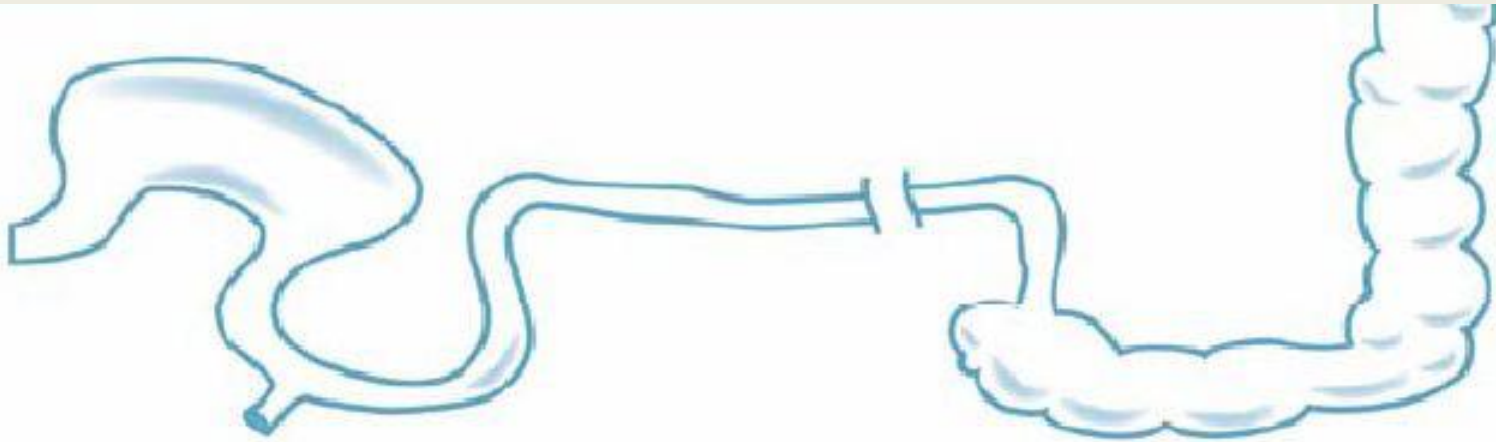
Stomach



organ	length (m)	surface (m ²)	pH	retention time	micro-organisms
stomach gaster	0.2	0.2	1.0-2.5	1-5 hrs	~10 ²

pH values

GI tract



Stomach

Jejunum

Ileum

Colon

pH

1.4-2.1

4.4-6.6

6.8-8.6

5-8

fasting

3-7

5.2-6.2

6.8-8.0

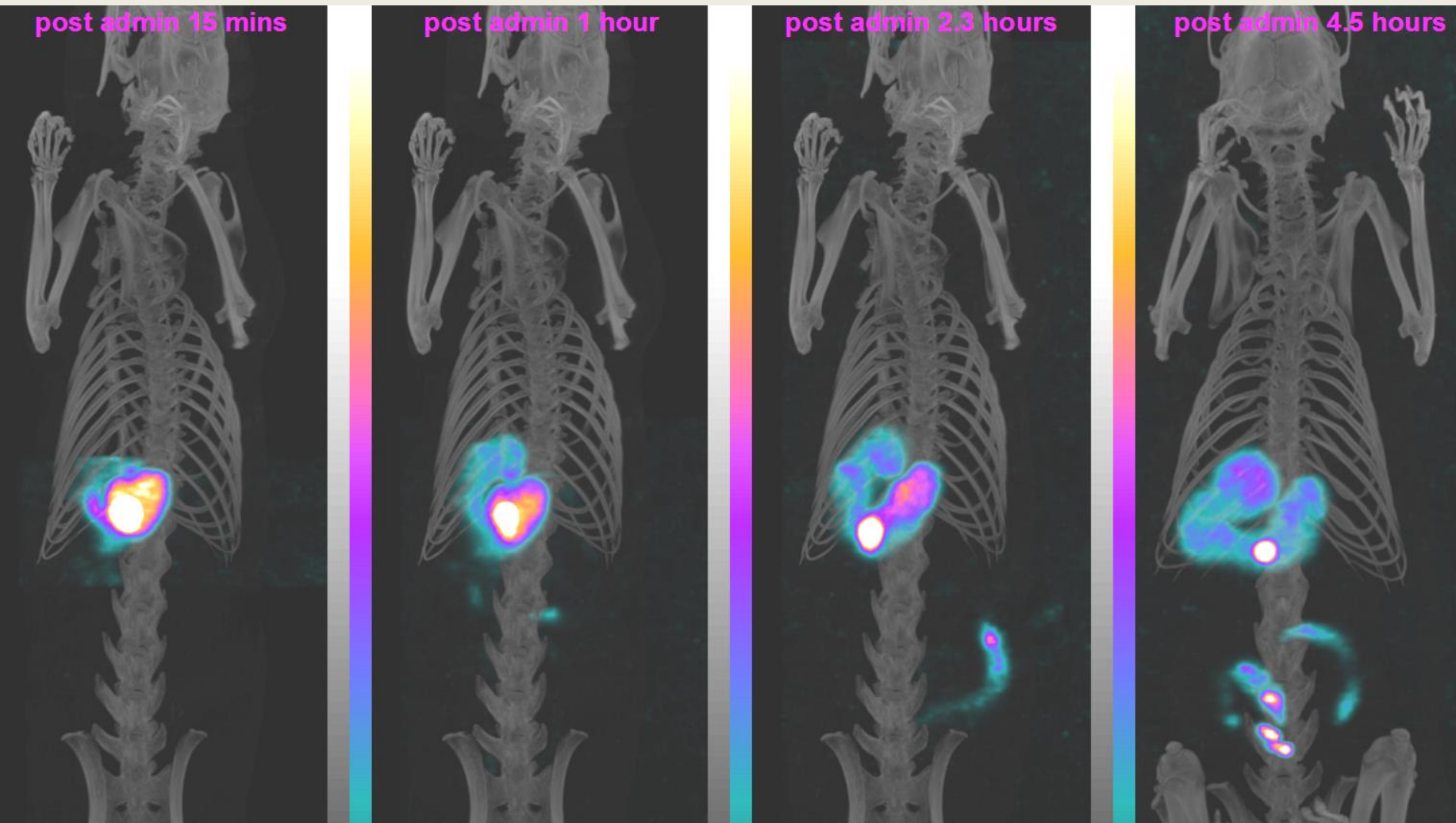
5-8

fed

FDDS – floating drug delivery system (Gastroretentive systems)



Gallium-76 isotope marked floating tablets



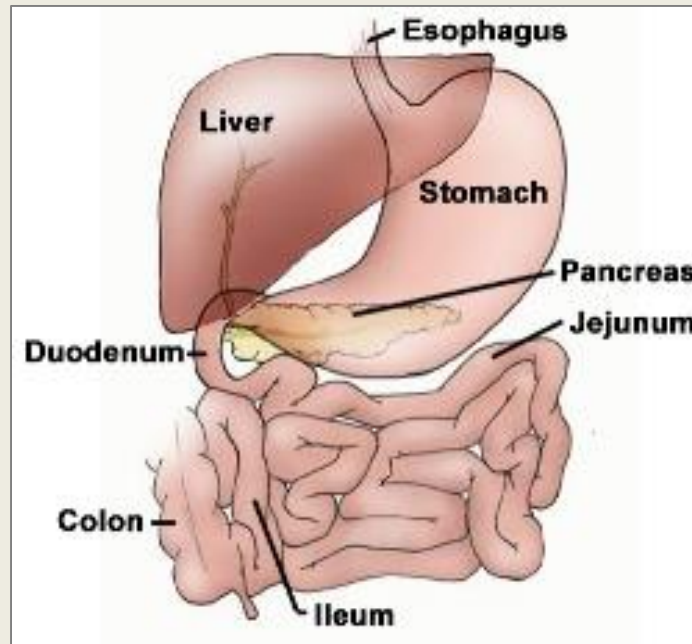
Preformulation studies and optimization of sodium alginate based floating drug delivery system for eradication of *Helicobacter pylori*

Péter Diós^{a, .}, Sándor Nagy^a, Szilárd Pál^a, Tivadar Perneckera^a, Béla Kocsis^b, Ferenc Budán^{c, d}, Ildikó Horváth^c, Krisztián Szigeti^c, Kata Bölcskei^e, Domokos Máthé^c, Attila Dévay^a

Absorption from intestines (bowel)

- Excellent blood and lymphatic circulation
- pH increases (3-7)
- Mostly weak bases can absorb from intestines
- The absorption mostly ends at the upper part of the jejunum
- Passive diffusion is the most important type of transport which is determined by the dissociation constant and the environment's pH

Small intestine

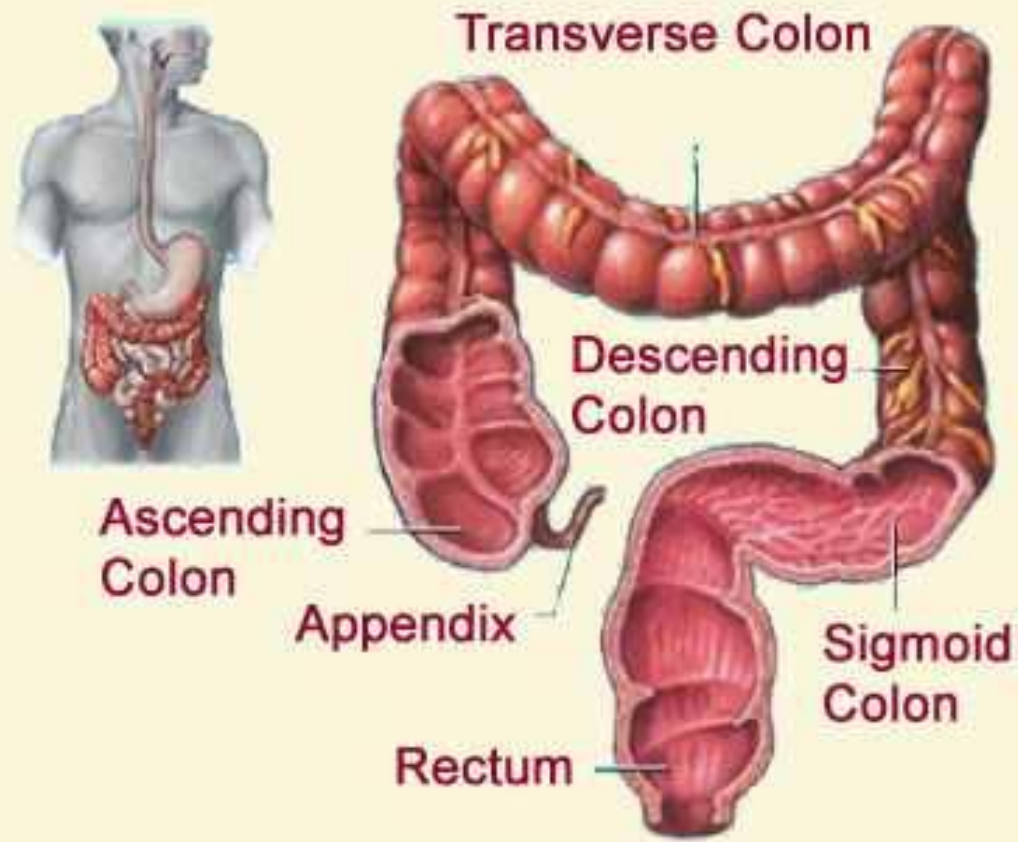


organ	length (m)	surface (m ²)	pH	retention time	micro-organisms
duodenum	0.3	0.02	5-6.5	>5 min	$\sim 10^2$
jejunum	3	100	6.9	1-2 hrs	$\sim 10^2$
ileum	4	100	7.6	2-3 hrs	$\sim 10^7$

Absorption from the small intestine

- Absorption occurs here in almost all cases
- Its surface is only 1 m², but actually it is huge because of the microvilli (100 m²)
- Duration of passage can influence the absorption (diarrhea, constipation)
- Short transit time decreases, long transit increases the absorption.

Large intestine

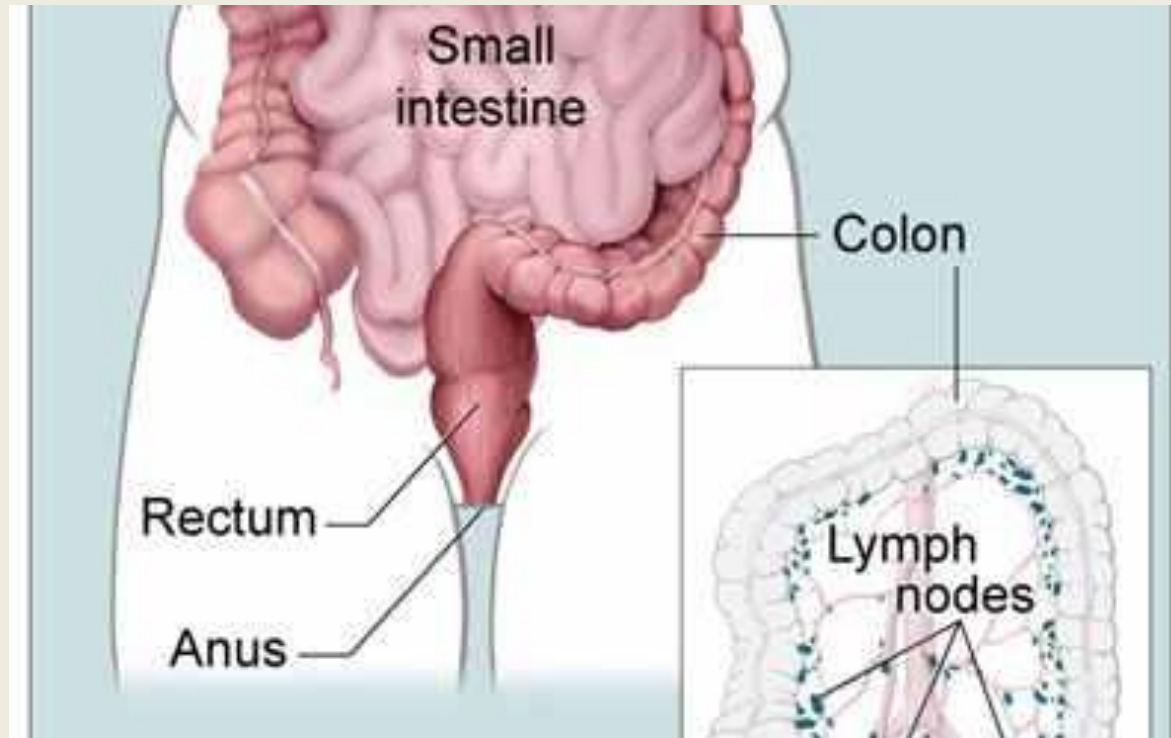


organ	length (m)	surface (m ²)	pH	retention time	micro-organisms
Large intestine cecum, colon	1.5	3	5.5-7.8	15-48 hrs	$\sim 10^{11}$

Absorption from the large intestine

- Absorption occurs in a negligible quantity
- It has its own bacterial flora, which can be utilized to achieve a local therapy of the colon

Rectum



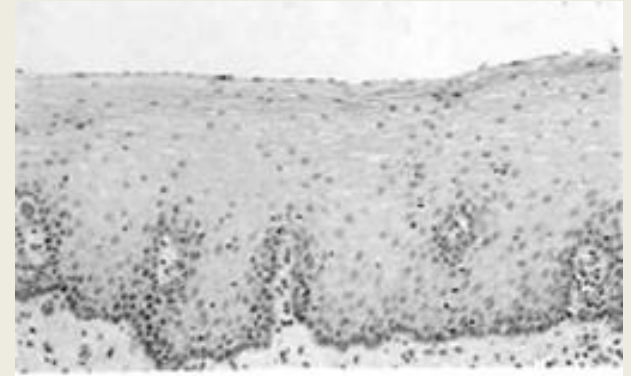
organ	length (m)	surface (m ²)	pH	retention time	micro-organisms
rectum	0,15-0,18	0,03	7,3-7,7	10-30 perc	~10 ¹⁰

Absorption from the rectum

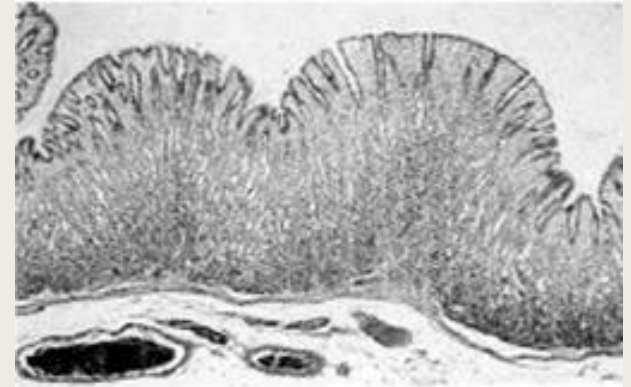
- Quick absorption
- No first-pass-effect
- In case of unconscious people and children
- In case substances that irritate stomach or intestines
- Suppository, enema

Absorption surfaces

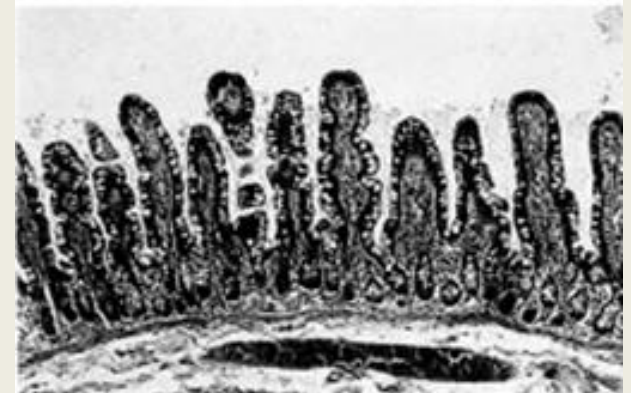
1. Mouth



2. Stomach



3. Small intestine

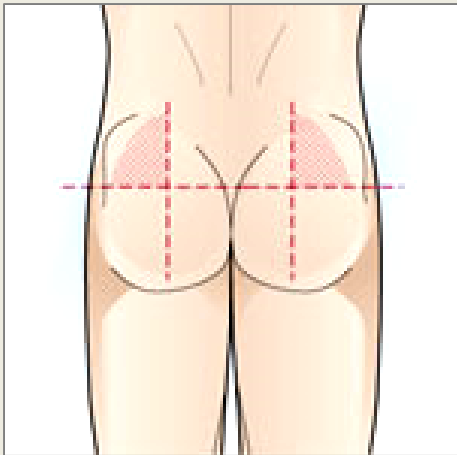


Parenteral absorption

Subcutaneous route

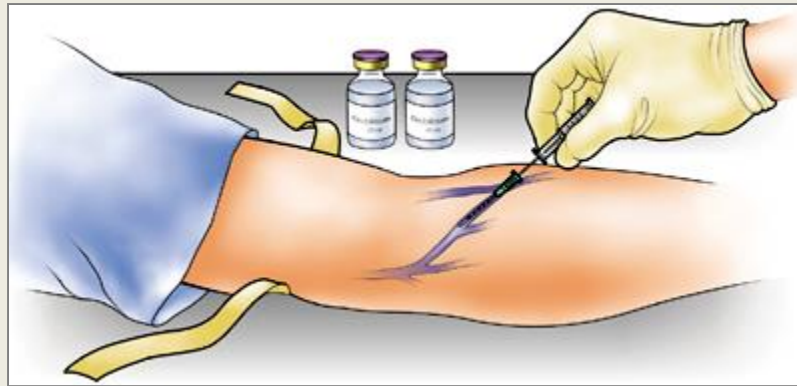


Intramuscular route



~~Absorption~~

Intravenous route



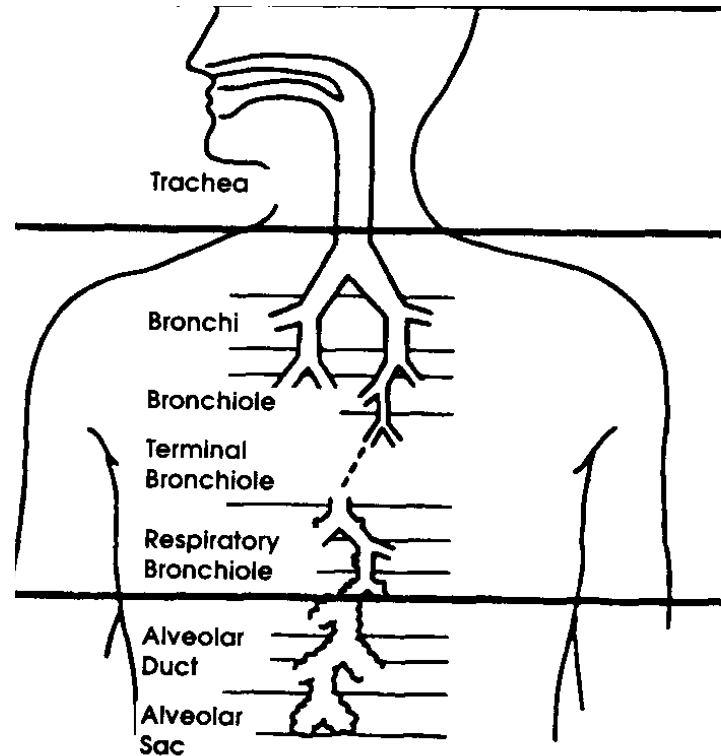
Parenteral absorption

- **Respiratory system**

– nasopharyngeal

– tracheobronchial

– pulmonary region



Absorption from the respiratory system

- Characteristics

- Alveolar region

- huge surface (150 m²)
 - 1-2 cell thickness – thin layer of cells

Absorption through the respiratory system

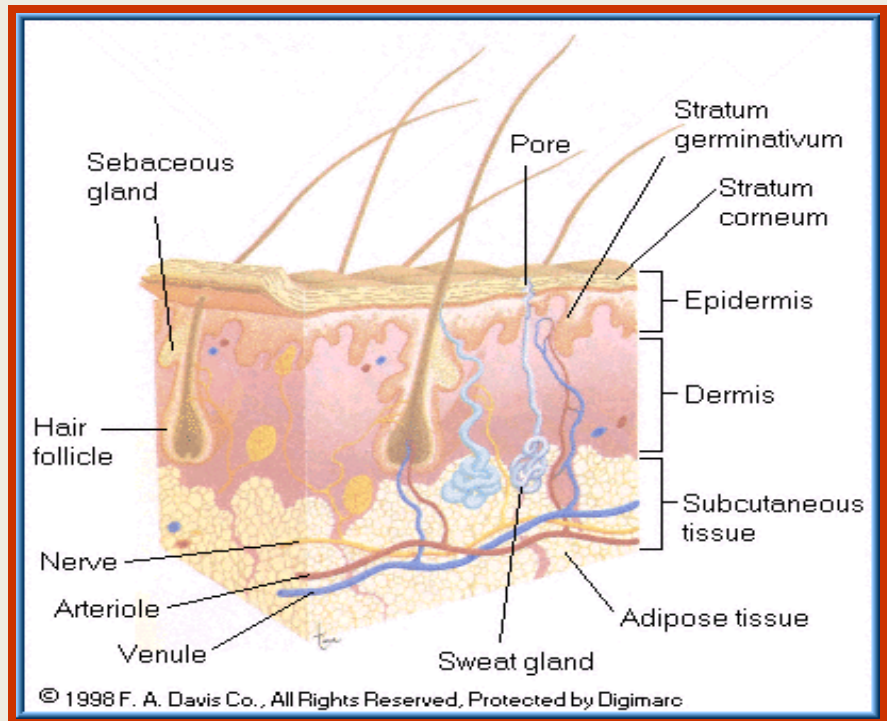
- Effect of size on the settling of particles
 - Nasopharyngeal 5-30 μm
 - Tracheobronchial 1-5 μm
 - Pulmonary 1-2 μm

Absorption through the respiratory system

- Absorption is determined by:
 - Ventillation
 - Concentration gradient
 - Alveolar surface
 - Alveolar capillaries' blood circulation
- Used in:
 - Inhalational anaesthesia (ether, halothane)

Absorption through the skin

- **Structure of the skin**
 - Epidermis
 - Dermis
 - Subcutaneous tissue



Thank you for your attention!